

Available online at www.sciencedirect.com

Tetrahedron Letters 45 (2004) 7479–7482

Tetrahedron Letters

A synthesis of 17-epi-calcidiol

Alicja Kurek-Tyrlik, Karol Michalak, Zofia Urbanczyk-Lipkowska and Jerzy Wicha*

Institute of Organic Chemistry, Polish Academy of Sciences, ul. Kasprzaka 44, 01-224 Warsaw, Poland

Received 18 June 2004; revised 27 July 2004; accepted 6 August 2004 Available online 26 August 2004

Abstract—The first synthetic approach to 17-*epi*-calcidiol 4 and congeners is presented. Key steps of the synthesis involve Pd-catalyzed reaction of the androst-16-ene derivative 6 with alkyl diazoacetates producing the respective cyclopropane derivatives 5, and lithium in liquid ammonia reduction of 5 leading to 17a-pregnane-20-carboxylic acid derivatives 9. The side chain was attached to ester 19 in a known manner.

2004 Elsevier Ltd. All rights reserved.

Calcitriol 1 (Fig. 1), the hormonally active metabolite of calciol (vitamin D_3), shows a broad spectrum of biolog-ical activities.^{[1](#page-3-0)} Its prime function consists in providing adequate calcium and phosphorus concentration in circulating fluids to support skeletal mineralization. Noncalcemic activities of 1 proved useful in controlling various human diseases, including skin and immune system metabolic disorders. Calcidiol 2 is the biosynthetic precursor of calcitriol showing similar but weaker biological activities. Several synthetic analogues of calcitriol are presently available as drugs or have been accepted for advanced clinical trials^{[2](#page-3-0)} and have stimulated the search for new structures having improved therapeutic properties.[3](#page-3-0) In continuing our work on synthesis and biological evaluation of calcitriol stereoisomers^{[4](#page-3-0)} we were interested in epimers of the natural products, 17-epi-calcitriol 3 and 17-epi-calcidiol 4. It was reasoned that 4 will be accessible from alkyl $16\alpha.20$ -cyclo-17 α -pregnane-21-carboxylates 5 by reductive fission of the C16–C20 bond, followed by side chain elaboration. Intermediates 5, in turn, would be obtained by diastereoselective cyclopropanation of the easily available androstane derivative 6. In the present communication we report the first synthesis of 17-*epi*-calciol derivatives and, en route, the synthesis of some 17-epi-cholesterol derivatives.

The starting material for the synthesis, 6, was prepared by reduction of 17-iodo-3 α , [5](#page-3-0)-cyclo-5 α -androst-16-ene⁵ with sodium in ethanol. To a solution of 6 in

Figure 1. Structures of calcitriol (1 α ,25-dihydroxy vitamin D₃) 1, calcidiol 2, their 17-epi-analogues 3 and 4, and the respective synthetic precursors.

Keywords: Vitamin D synthesis; Cyclopropanation; Alkyl diazoacetates; Lithium in liquid ammonia.

^{*} Corresponding author. Tel.: +48 22 6328117; fax: +48 22 632 6681; e-mail: jwicha@icho.edu.pl

^{0040-4039/\$ -} see front matter © 2004 Elsevier Ltd. All rights reserved. doi:10.1016/j.tetlet.2004.08.050

dichloromethane (DCM) containing palladium(II) acetate, 6 a solution of *tert*-butyl diazoacetate in DCM was slowly added^{[7](#page-3-0)} by means of a syringe pump. A product was obtained to which structure 5a (Scheme 1) was assigned on the basis of its ${}^{1}H$ and ${}^{13}C$ NMR spectra. Product 5a was contaminated with side products derived from the diazoester (ca. 5%) and could not be purified by column chromatography. Treatment of 5a with sodium methoxide in methanol at reflux followed by chromatography afforded pure methyl ester 5d in a 93% yield from 6.

Palladium-catalyzed reaction of 6 with ethyl diazoacetate (commercial reagent) provided a mixture of cyclopropane derivatives 5b and 5c differing in configuration at C20, in a ratio of ca. 7:1 (by $1H$) NMR). This mixture was subjected to methanolysis to give a single isomer 5d in 89% yield from 6. Similarly, 6 and methyl diazoacetate gave 5d and 5e in a ratio of 7:1. Treatment of tert-butyl ester 5a (ca. 90% pure) with lithium in liquid ammonia $8-10$ afforded a mixture of products. A neutral fraction, isolated in ca. 50% yield, was reduced with $LiAlH₄$ to give a mixture of alcohols that were identified as $7(36\% \text{ yield})$ and $10(11\% \text{ yield})$. The acid fraction contained mainly compound 12. No product of alternative fission of the C17–C20 bond was detected. It was subsequently found that reduction of methyl (5d) or ethyl esters (mixture 5b and 5c) also proceeded virtually regioselectively; moreover with these esters only small amounts of acid 12 were formed.^{[11](#page-3-0)} In the most efficient procedure developed (Scheme 1), methyl ester 5d was treated with Li in liquid ammonia containing tert-butyl alcohol and THF at reflux temperature to give a mixture of alcohol 7, aldehyde 8 and alcohol 10 with the cyclopropane ring retained, in nearly quantitative yield. Since chromatographic separation of the undesired component of the mixture (10) was problematic, the whole reduction product was oxidized^{[12](#page-3-0)} with DMSO in the presence of $\overline{P}y \cdot SO_3$ and the resulting mixture of aldehydes was further oxidized with potassium permanganate in tert-butanol in the presence of $NaH₂PO₄$ to give a mixture of carboxylic acids 9 and 12. At this stage, acid 9 was easily purified by chromatography on silica gel. Crystalline 9 was obtained in 76% overall yield from 5d.

Alkylation of methyl ester 13 (obtained by esterification of 9 with diazomethane) was carried out via an earlier developed method[.13](#page-3-0) It was rewarding to learn that the reaction affords the diastereomerically pure product (as an oil) in nearly quantitative yield. Several derivatives of this product were examined in the search for a crystalline material suitable for single crystal X-ray analy-sis. The structure was eventually solved^{[14](#page-3-0)} for acid 17 prepared as showed in [Scheme 2.](#page-2-0) It revealed that the product 14 with the (20S) configuration was formed. X-ray analysis also confirmed the assumed stereoselectivity in the cyclopropanation reaction and the regioselectivity of the reductive cleavage of the cyclopropane ring [\(Fig. 2\)](#page-2-0).

The side chain was attached to 14 in a known manner, ¹⁵ as illustrated in [Scheme 2](#page-2-0). Ester 14 was reduced to alcohol 15. The respective tosylate 16 was allowed to react with the lithium derivative of 3-methyl-1-butyn-3-yl 2-tetrahydropyranyl ether, and the product 20 was subjected to catalytic reduction followed by removal of the protective groups. 25-Hydroxy-17-epi-cholesterol 18 was obtained in over 70% yield from 14.

Acetate 19 was subjected to a bromination–dehydrobro-mination procedure^{[16](#page-3-0)} to give 17-*epi*-7-dehydrocholesteryl acetate 22 in ca. 40% yield. This product was hydrolyzed and the crude diol 23 was dissolved in a mix-

Scheme 1. 16a,17a-Cyclopregnan-20-oic acid derivatives obtained by palladium-catalyzed reaction of ene 6 with alkyl diazoacetates, and their reduction products.

Scheme 2. Synthesis of the 17-epi-cholesterol derivatives 18 and 19.

Figure 2. ORTEP plot of the X-ray structure of 17.

ture of diethyl ether and toluene, and irradiated with a medium-pressure mercury lamp until ca. 50% of the substrate was consumed (HPLC). Then, the solvent was removed under reduced pressure and the crude product was dissolved in ethanol and heated under reflux for 6 h. The resulting product was chromatographed on a preparative HPLC column to give 17-epi-calcidiol 4 (Scheme 3).

The transformation of 4 into 3 and the biological activity of these compounds will be reported in due course.

Spectroscopic and analytical data of selected new compounds: Methyl $3\alpha, 5:16\alpha:20(S)$ -bicyclo-5 α ,17 α -pregnane-21-oate (5d) ¹H NMR (200 MHz, CDCl₃): 0.41 (dd, $J = 8.1$, 5.1 Hz, 1H, C4–Ha), 0.62 (t, $J = 5.1$ Hz, 1H, C4–Hb), overlapping 0.5–0.7 (m, 1H), 0.90 (s, 3H, C18–H), 1.00 (s, 3H, C19–H) overlapping 0.7–1.9 (m, 18H) 2.73 (t, $J = 2.6$ Hz, 1H, C6–H), 3.29 (s, 3H, OMe), 3.6 (s, 3H, COOMe); ¹³C NMR (50MHz): 13.02, 19.20, 19.66, 20.13, 21.32, 22.37, 24.82, 25.07, 27.04, 28.98, 33.16, 35.08, 35.20, 35.31, 38.32, 41.07, 43.51, 47.16, 48.61, 51.41, 56.53, 82.09, 174.58; HRMS: calcd for $C_{23}H_{34}O_3$: 358.25080; found: 358.25175.

Methyl 6β -methoxy-3 α , 5-cyclo-23, 24-bisnor-5 α , 17 α cholane $(14)^{11}$ H NMR $(200 \text{ MHz}, \text{ CDCl}_3)$: 0.42 $(\text{dd},$ $J = 8.1, 5.1$ Hz, 1H), 0.63 (t, $J = 4.9$ Hz, 1H,), 0.84 (s, 3H, C-19H), 1.00 (s, 3H, C-18H), 1.09 (d, $J = 6.8$ Hz, 3H, C–21H) overlapping, 0.7–2.1 (m, 19H), 2.28 (dq, $J = 10.5, 6.8$ Hz, 1H, C-20H), 2.76 (t, $J = 2.8$ Hz, 1H, C–6H), 3.31 (s, 3H, OMe), 3.65 (s, 3H, COOMe); 13 C NMR: 13.77, 17.88, 19.90, 22.12, 23.45, 25.59, 26.44, 26.49, 31.26, 34.00, 34.75, 35.86, 42.09, 36.29, 44.04, 44.32, 48.37, 51.07, 52.01, 52.62, 57.15, 82.94, 178.6; HRMS: calcd for C₂₄H₃₈O₃: 374.28210; found: 374.28173.

Scheme 3. Synthesis of 17-epi-calcidiol 4.

 17α -Cholesteryl acetate (19): ¹H NMR (500 MHz, CDCl₃) 0.77 (s, 3H, C-18H), 0.82 (d, $J = 6.5$ Hz, 3H, C–21H), 1.02 (s, 3H, C–19H) and 1.21 (s, 6H, C–26 and C–27H) overlapping 1.10–1.80 (m, ca. 24H), 1.80– 1.90 (m, 2H), 2.03 (s, 3H, COMe), 2.29–2.35 (m, 2H), 4.56–4.64 (m, 1H, C–3H), 5.36–5.39 (m, 1H, C–5H), 13C (125MHz) 16.98, 19.29, 21.07, 21.04, 22.12, 22.27, 26.28, 27.79, 29.26, 29.27, 32.33, 32.45, 33.01, 38.12, 38.37, 43.52, 44.24, 50.13, 52.65, 52.85, 71.03, 73.97, 122.63, 139.61, 170.49. Elemental analysis: calcd for $C_{29}H_{48}O_3$ (444.69): C, 78.33, H, 10.88%; found: C, 78.37, H, 10.74%.

(5Z,7E)-3S-9,10-seco-17a-Cholesta-5,7,10(19)-trien-3b-ol (4): *k* (EtOH) 264 nm; ¹ H NMR (500MHz, CDCl3) 0.63 (s, 3H, C-18H), 0.86 (d, $J = 6.6$ Hz, 3H, C-21H), 1.21 (s, 6H, C–26 and C–27H) and 1.54 (s, H₂O) overlapping 1.05–1.75 (m, ca. 24H), 1.89–1.96 (m, 1H), 2.11–2.21 (m, 2H), 2.25–2.31 (m, 1H), 2.55–2.59 (m, 1H), 2.81–2.85 (m, 1H), 3.90–3.97 (m, 1H, C–3H), 4.82 (d, $J = 2.6$ Hz, 1H, C-19HE), 5.05 (dt, $J = 1.3$, 2.6Hz, 1H, C-19HZ), 6.05 (dt, $J = ca$, 1, 11.3Hz, 1H, C–7H), 6.23 (d, $J = 11.2$ Hz, 1H, C–6H); EI MS 400 (26) M⁺, 382 (19) (M-18)⁺, 367 (20) (M-18-15), 349 (17) , 271 (18) (M-side chain), 253 (18) $(217 - H₂O)$, 211 (15), 176 (22), 161 (18), 158 (60), 136 (100) (C1– C8 fragment), 118 (98) ($136-H₂O$); HR MS calcd for $C_{27}H_{44}O_2$: 400.3341; found: 400.3338.

References and notes

- 1. For leading references, see: (a) Bouillon, R.; Okamura, W. H.; Norman, A. W. Endoc. Rev. 1995, 16, 200–257; (b) Beckman, M. J.; DeLuca, H. F. Modern view of vitamin D3 and its medicinal uses. In Progr. Med. Chem.; Ellis, G. P., Luscombe, D. K., Oxford, A. W., Eds.; Elsevier: Amesterdam, 1998; Vol. 35, pp 1–56.
- 2. For a recent review, see: Posner, G. H.; Kahraman, M. Eur. J. Org. Chem. 2003, 3889–3895; Posner, G. H.; Kahraman, M. Eur. J. Org. Chem. 2003, 4937–4937.
- 3. Recent works include: (a) Sussman, F.; Rumbo, A.; Villaverde, M. C.; Mourino, A. J. Med. Chem. 2004, 47, 1613–1616; (b) Chen, Y. J.; Gao, L. J.; Murad, I.; Verstuyf, A.; Verlinden, L.; Verboven, C.; Bouillon, R.; Viterbo, D.; Milanesio, M.; Van Haver, D.; Vandewalle, M.; De Clercq, P. J. Org. Biomol. Chem. 2003, 1, 257–267; (c) Fall, Y.; Diouf, O.; Gomez, G.; Bolano, T. Tetrahedron Lett. 2003, 44, 6069–6072.
- 4. (a) Marczak, S.; Przezdziecka, A.; Wicha, J.; Steinmeyer, A.; Zuegel, U. Bioorg. Med. Chem. Lett. 2001, 11, 63–66; (b) Przezdziecka, A.; Achmatowicz, B.; Marczak, S.; Steinmayer, A.; Wicha, J.; Zuegel, U. Synthesis of the Enantiomer of Natural 1 α ,25-Dihydroxy Vitamin D₃ and Related Stereoisomers. In Vitamin D Endocrine System. Proceedings of the 11th Workshop on Vitamin D; Norman, A. W.; Bouillon, R.; Thomasset, M., Eds.; University of California, Riverside (Calif.), 2000; pp 81–84.
- 5. Przezdziecka, A.; Kurek-Tyrlik, A.; Wicha, J. Collect. Czech. Chem. Commun. 2002, 67, 1658–1668.
- 6. (a) Anciaux, A. J.; Hubert, A. J.; Noels, A. F.; Petiniot, N.; Teyssie, P. J. Org. Chem. 1980, 45, 695–702; (b) Majchrzak, M. W.; Kotełko, A.; Lombert, J. B. Synthesis 1983, 469–470.
- 7. Doyle, M. P. Chem. Rev. 1986, 86, 919–939.
- 8. House, H. O.; Blankley, C. J. J. Org. Chem. 1968, 33, 47– 53.
- 9. For a review, see: von Angerer, S. Cyclopropanes: Transformations, Ring-opening Reactions—by Reductive Fission. In Carbocyclic Three-Membered Ring Compounds; de Meijere, A., Ed.; Georg Thieme: Stuttgart, 1996; Vol. E17c, pp 2045–2053.
- 10. For recent work, see: Srikrishna, A.; Ramasastry, S. S. V. Tetrahedron Lett. 2004, 45, 379–382.
- 11. With ester 5a cleavage of the C–O bond (path b) is preferred over cleavage of the C–C bond (path a), presumably due to the stability of the tert-butyl radical.

- 12. Parikh, J. R.; Doering, W. v. E. J. Am. Chem. Soc. 1967, 89, 5505–5507.
- 13. Wicha, J.; Bal, K. J. Chem. Soc., Perkin Trans. 1 1978, 1282–1288.
- 14. Crystallographic data (excluding structure factors) for the structure in this paper have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication numbers CCDC 242072. Copies of the data can be obtained, free of charge, on application to CCDC, 12 Union Road, Cambridge BC2 1EZ, UK [fax. +44-(0)1223-336033 or e-mail:deposit@ccdc.cam.ac.uk].
- 15. Partridge, J. J.; Faber, S.; Uskokovic, M. Helv. Chim. Acta 1974, 57, 764–770.
- 16. For an example, see: DeLuca, H. F.; Wicha, J. Method of Treating Hypoparathroidism with 20(S) Vitamin D Compounds. U.S. Patent 5,552,392; Appl. November 3, 1993, 1996.